

Lines within a band are characterized by:

$$\nu = c + dm + em^2$$

$c, d \& e$ are constants.

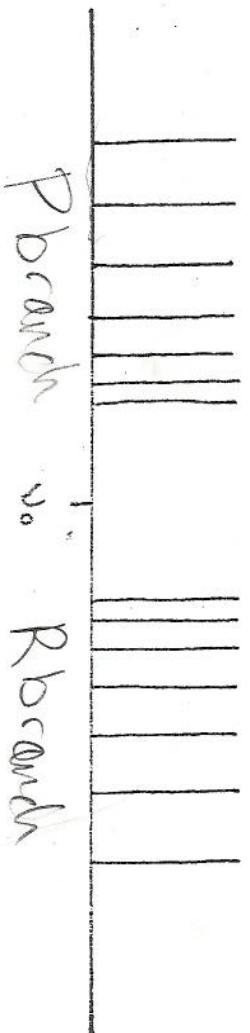
m can be positive or negative

$a + m = 0$; $\nu = c$ this corresponds to a missing line between $m = +1$ and $m = -1$.

This wavenumber is called the null gap = ν_0 .

$m < 0$

$m > 0$



Series of lines with $m > 0$ is the R branch $\Delta J = +1$

Series of lines with $m < 0$ is the P branch $\Delta J = -1$

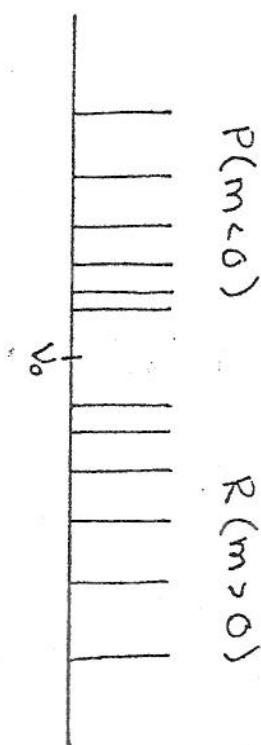
P & R branches go in different directions

the sign of e is determined by the way the P & R lines draw apart or together.

$P(m<0)$ $R(m>0)$



e is negative



$R(m>0)$

e is positive

* However this depends heavily on d & e terms because m^2 term can dominate and cause the lines to bend back toward the null gap.

$$\nu = \nu_0 + dm + em^2$$

For the CH spectrum, e is negative and the R band reverses after $m=7$, back toward the null gap.

Since there are different values for B_v from one vibrational state to another the observed spectra corresponds to transitions, not particular states.

The lines are then represented by

$$\nu_R = \nu_0 + 2B'_v + (3B'_v - B''_v)J + (B'_v - B''_v)J^2; J=0, 1, 2, \dots$$

$$\nu_P = \nu_0 - (B'_v + B''_v)J + (B'_v - B''_v)J^2; J=1, 2, 3.$$

$$\nu_Q = \nu_0 + (B'_v - B''_v)J + (B'_v - B''_v)J^2; J=1, 2, 3, \dots$$

R branch $m = -(J+1)$

P branch $m = +J$

Q branch $m = J$

$\Delta J = 0, \pm 1$ and lowest J value is $J=1$ because $J < n$

1st line of : $R \rightarrow J=0$

$P \rightarrow J=2$

$Q \rightarrow J=1$.

$P(J=1)$ line is missing when a Q branch is present.

Data Reduction

from recorded 2nd order spectra

1) Find null gap:

$$\nu_0 \text{ is usually between } P(1) \text{ and } R(1)$$

if $P(1)$ is missing use the Q branch
because it often runs right up to
null gap depending on the $(B_v' - B_v'')$

$$\nu_Q = \nu_0 + (B_v' - B_v'') J(J+1)$$

so $J=1$ line would be close to ν_0
if $(B_v' - B_v'')$ is small, so look
for crowded region.

If $(B_v' - B_v'')$ is large then there will
be a region with no peaks.

There is a graphical method, however it
requires B_v values - so it time consuming
iterative procedure.

also the P & P densities must obey combination relations separately.

$$R(J-1) - R(J)$$

$$\Delta_2 F' = R(J) - R(J)$$

$$\Delta_2 F = 4 B_v (J + \frac{1}{2}),$$

$$\Delta_2 F'' = R(J-1) - R(J+1)$$

J	$\frac{\Delta_2 F'}{4(J + \frac{1}{2})}$	$\frac{\Delta_2 F''}{4(J + \frac{1}{2})}$
0		
1	12.13	12.73
2	10.93	12.35
3	10.37	11.71
4	10.0	11.4
5	9.76	11.19
6	9.67	10.88
7	9.43	10.81
8	9.33	10.67
9	9.21	10.56
10	9.07	10.50
11	8.97	10.44
12	8.83	10.3
13		10.19
14		

$$\langle B_v' \rangle = 9.81$$

$$\langle B_v'' \rangle = 11.05$$

Finding final average of combination relations;

$$\langle B_v' \rangle_f = \frac{386.06}{40} = 9.65$$

$$\langle B_v'' \rangle_f = \frac{465.57}{42} = 11.09$$

$$\text{From } (B_v' - B_v'') = -1.44$$

$$B_v' + B_v'' = 20.74$$

$$\text{From the above values, } m_{\text{mean}} = \frac{(B_v' + B_v'')}{-2(B_v' - B_v'')} = 7.2 \text{ which is in very good agreement.}$$

From this technique :

$$\langle B_v' \rangle = 9.65 \quad (12.18)$$

$$\langle B_v'' \rangle = 11.09 \quad (13.81).$$

$$B_v' - B_v'' = -1.44 \quad (-1.635)$$

$$B_v' + B_v'' = 20.74 \quad (26.0).$$

check

$$m_{\text{head}} = \frac{B_v' + B_v''}{2(B_v' - B_v'')} = 7.2$$

5) Draw Fortrat Diagram - using wavenumbers in air

6) Determine wavenumber relation:

$$v_{p,R} = v_0 + (B_v' + B_v'')m + (B_v' - B_v'')m^2$$

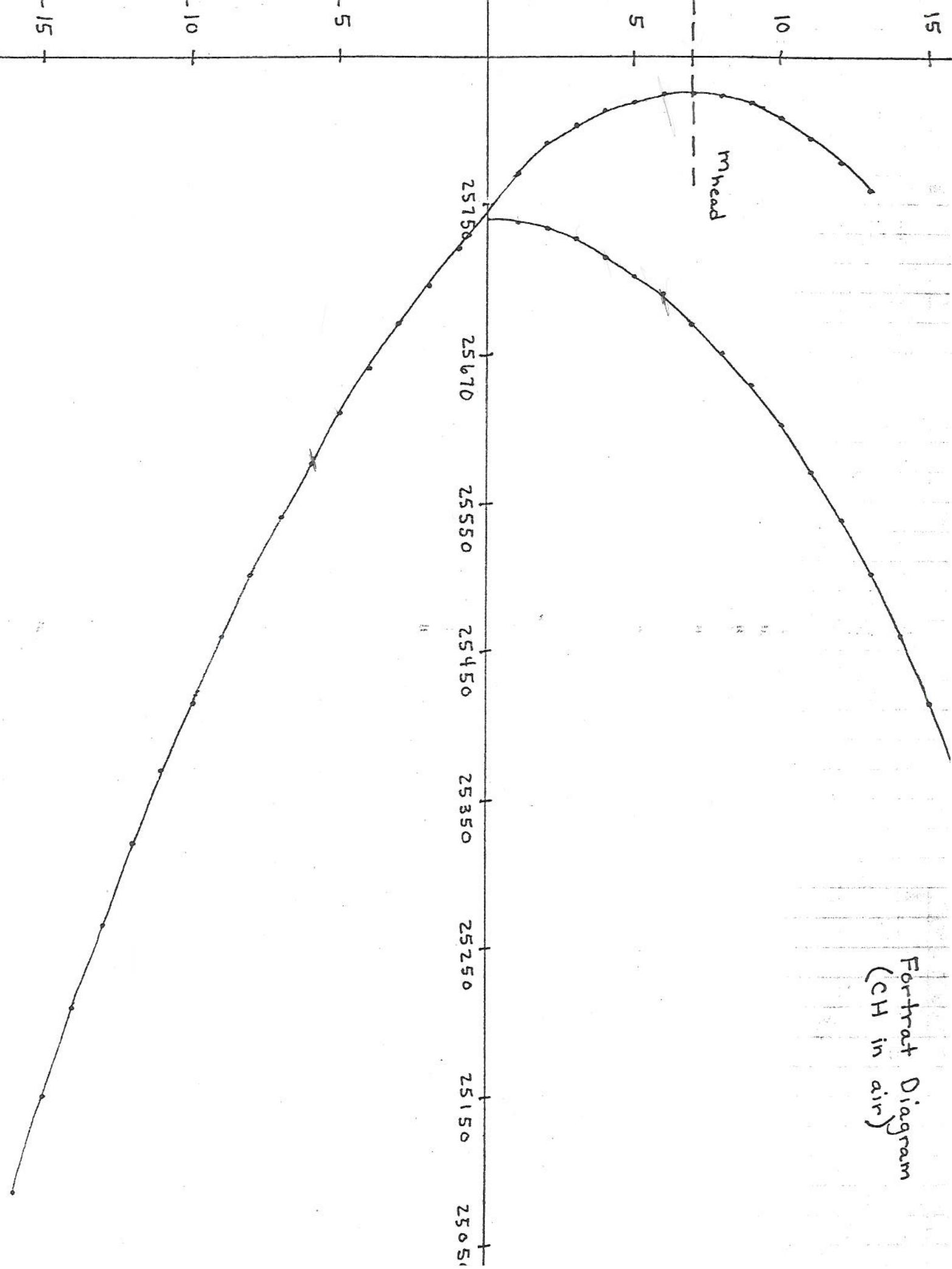
$$v_Q = v_0 + (B_v' - B_v'')J(J+1)$$

Results :

$$v_{p,R} = 25731.5 + 20.74m - 1.44m^2$$

$$v_Q = 25731.5 - 1.44 J(J+1)$$

Fortrag Diagram
(CH in air)



where ν_0 has been corrected to vacuum value:

in order to use this formula, the peak wavenumbers must be converted from air to vacuum values.

$$\nu_{\text{vac.}} = \nu_{\text{air}} \cdot \frac{n_{\text{vac.}}}{n_{\text{air}}}.$$

for the wavelength range 3850 \AA to 4100 \AA index of refraction

$$n_{\text{air}} \approx 1.0003 \quad \text{approximately constant}$$

* is was irrelevant when determining B_v values because used differences.

1) Draw Fortrat Diagram - with vacuum corrected experimental values and with calculated values from correlation.

$$\nu_{P,R} = 25745.0 + 20.74 m - 1.44 m^2$$

$$\nu_Q = 25745.0 - 1.44 J(J+1)$$

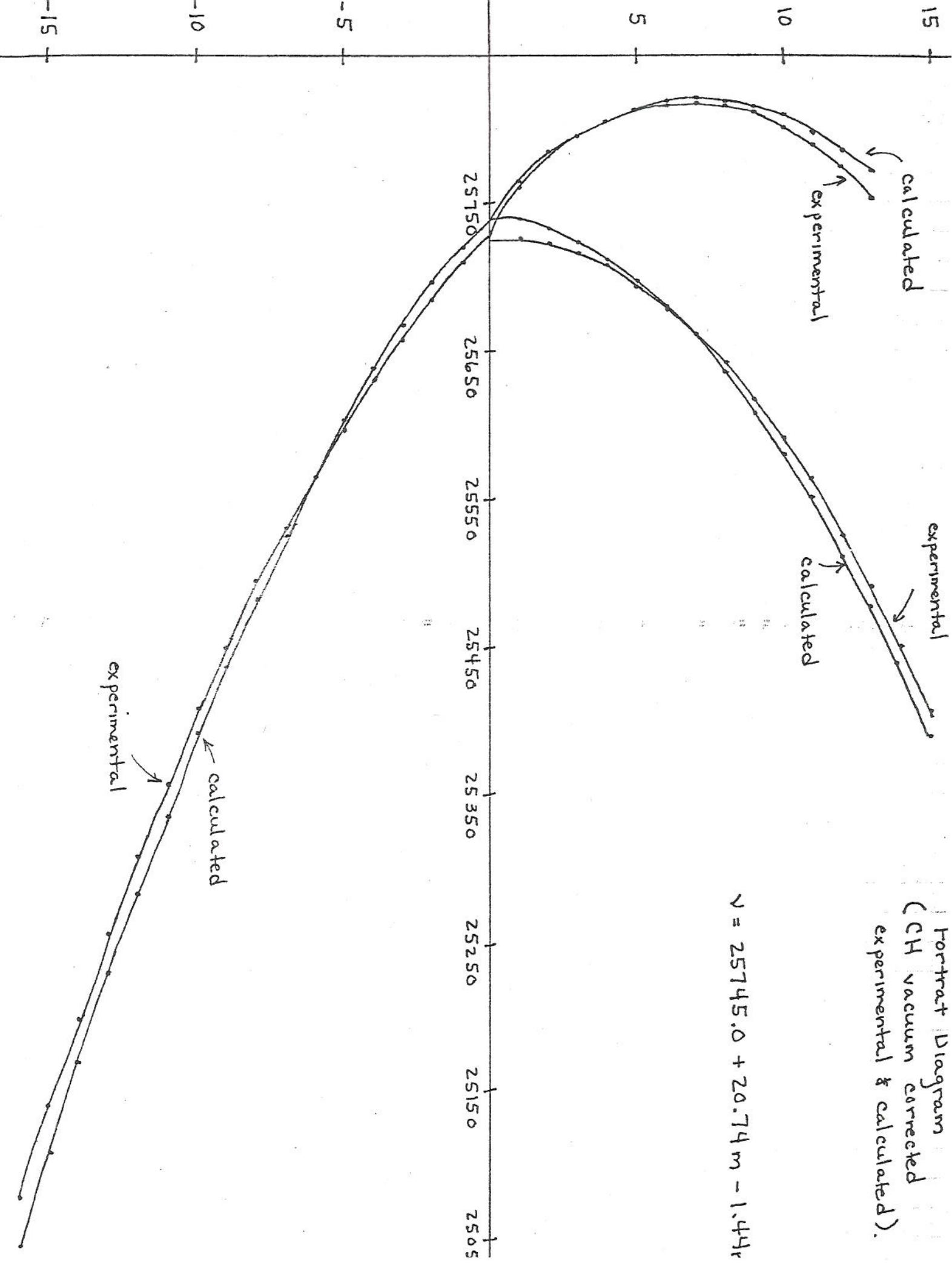
have determined ν_0 null gap from $R(J=0)$ line & B_v' value

$$\nu_R(J=0) = \nu_0 + 2B_v' \quad B_v' = 9.65$$

$$\nu_0 = 25745.0$$

Fortrat Diagram
(CH vacuum corrected
experimental & calculated).

$$\nu = 25745.0 + 20.74m - 1.44r$$



* this is possible because all wavenumber values are based on
 $R(J=6)$ head at 3872\AA , not the null gap.

8) Determine internuclear spacing:

$$B_{v'} = Be - \alpha_e(u' + \frac{1}{2}) = 9.65$$

$$B_{v''} = Be - \alpha_e(u'' + \frac{1}{2}) = 11.09$$

the CH 3850\AA to 4020\AA band is a 0,0 transition

$$u' = u'' = 0$$

using α_e values from Herzberg:

$$\alpha_e(^2\Sigma^-) = .485$$

$$\alpha_e(^2\Pi) = .534$$

$$\alpha_e = \sqrt{\frac{8\pi^2\mu k_e e c}{h}}$$

$$Be' = 9.893 \quad \text{actual values } Be(^2\Sigma^-) = 12.89$$

$$Be'' = 11.357$$

$$Be(^2\Pi) = 14.46$$

using

$$B_e = \frac{h}{8\pi^2 c \mu r_e^2}$$

$$\mu = \frac{m_e m_H}{m_e + m_H}$$

$$r_e' = 1.35 \text{ \AA}$$

actual values

$$1.18 \text{ \AA}$$

$$r_e'' = 1.26 \text{ \AA}$$

$$1.12 \text{ \AA}$$

q) Rotational Temperature:

determine this from Boltzmann's Law about populations in different J states. - intensity relationships

however did not have any of the required information so used from Herzberg, for unresolved band lines.

$$\Delta V_{\max} (P+R) = \sqrt{\frac{8 B K T}{h c}}$$

$$R_{\max} (J=6) = 25826.4$$

$$P_{\max} (J=7) = 25541.6 \quad B_e = 10.625$$

$$T_{\text{rot.}} = 1373^\circ \text{K}$$

If $P_{\max} (J=10) = 25417.9$

$$T_{\text{rot.}} = 2825^\circ \text{K}$$

Recommendations :

- 1) take spectra in 2nd order.
- 2) use Broida spectrum & Herzberg book.
- 3) may have to do the spectrum more than once.
- 4) take a peak reference point rather than guess at
Vo.

Q lines are present when $\Lambda \neq 0$ $\Lambda = m_L$
 L = electronic orbital angular momentum.

For CH the transition is $^2\Sigma^- (\Lambda=0) \rightarrow ^2\Pi (\Lambda=1)$

$\Lambda \neq 0$ corresponds to another allowable transition

$$\Delta \Lambda = 0, \pm 1$$

Diatomic molecule can be represented by Vibrating Rotator

because vibrations & rotations occur simultaneously

define

$$B_v = \frac{\hbar}{8\pi^2 c \mu} \left[\frac{1}{r^2} \right]$$

where

$$B_v = B_e - \alpha e (v + \frac{1}{2})$$

v = vibration state

so that vibration & rotation terms are combined in an average approximation.

* have neglected higher order terms.

Results:

using the Broida spectra determined the peak species and m numbers. - too complicated

- i) Q branch
- ii) have A splitting and spin splitting in Q, P and R branches.
- iii) Have R branch bending back toward null gap.

using the $P(2)$ and $R(1)$ lines made ν_0 estimate;

$$\nu_0 = 25726.8 \text{ cm}^{-1}$$

* using the $R(7)$ head at 3872\AA as reference to determine wavelengths.

from this data $(B_u' - B_u'') \sim -1.6$ which is small, so $Q(1)$ line should be very close to ν_0 .

A new null gap was placed at

$$\nu_0 = 25739.2$$

2) Compile wavenumbers of each peak & determine the 1st and 2nd differences of peaks.

$$\nu = \nu_0 + dm + em^2$$

$$\Delta\nu = \nu(m+1) - \nu(m)$$

$$\Delta^2\nu = \Delta\nu(m+1) - \Delta\nu(m)$$

should be relatively constant

$$\frac{\nu}{\nu_{corr.}} = \frac{\Delta\nu}{\Delta^2\nu}$$

ν_0	m	$\Delta\nu$	$\Delta^2\nu$
25814.8	4	25807.1	9.6
25805.2	3	25797.5	12.1
25793.1	2	25785.3	8.7
25772.3	1	25764.3	20.8
25739.3	0	25731.5	
-	-1		
25695.9	-2	25688.2	2.5
25669.6	-3	25661.9	2.1
25641.2	-4	25633.5	28.4

$$\nu_{vacuum} = \nu_{air} \cdot \frac{n_{vac.}}{n_{air}}$$

After examining Δv values between a P(2) line and R(1) line & using
 $R(m=7)$ head value at 2772 Å, $\nu_0 = 25728.6$.
& $\frac{B_1 - B_1''}{B_1 + B_1''} = -1.3$, which is small, so reassigned
 ν_0 for close to Q($J=1$) line.

Again using $P(m=7)$ head : $\nu_0 = 25739.2$
 $\lambda_0 = 3225.1$

<u>m</u>	<u>λ</u>	<u>ν_{nm}</u>	<u>Δv</u>	<u>$\Delta^2 v$</u>
13	14	too close to ν_0		
12	13	25761.6	21.3	
11	12	25782.9	16.0	5.3
10	11	25798.9	12.7	3.3
9	10	25811.6	7.5	5.2
8	9	25819.1	5.3	2.2
7	8	25824.4	2.0	3.3
6	7	25826.4	2.0	0
5	6	25824.4	3.2	1.2
4	5	25821.2	6.4	3.2
3	4	25814.8	9.6	3.2
2	3	25805.2	12.1	2.5
1	2	25793.1	20.2	8.7
0	1	25772.3		
0	0	25731.2	25731.5	
1	-1	(25720.3) ?	(25712.6) ?	
2	-2	25635.2	25622.2	
3	-3	25662.6	25661.9	2.1
4	-4	25641.2	25633.5	3.2
5	-5	25607.6	25601.9	2.9
6	-6	25575.1	25567.4	-1.0
7	-7	25541.6	25533.9	6.1
8	-8	25502.0	25494.3	.9
9	-9	25461.5	25453.9	3.1
10	-10	25417.9	25410.3	2.8
11	-11	25370.5	25362.9	2.9
12	-12	25322.2	25312.6	6.0
13	-13	25277.9	25260.2	1.0
14	-14	25210.3	25203.3	2.9
15	-15	25151.0	25143.5	3.7
16	-16	25087.4	25079.9	4.7
17	-17	25013.1	25011.6	4.7
				$\Delta^2 v_{RR}$
				$= 3.256$
				$(B_1' - B_1'') = 1.628$

<u>J</u>	<u>N₀₅₅</u>	<u>ΔV</u>	<u>Δ²V</u>
1	25737.2	25729.5	3.2
2	25734.0	25726.3	7.4
3	25726.6	25718.9	9.5
4	25717.1	25709.4	11.6
5	25705.5	25697.8	14.8
6	25690.7	25683.1	17.9
7	25672.8	25665.1	20.1
8	25652.7	25645.0	23.1
9	25629.6	25621.9	27.3
10	25602.3	25594.6	30.4
11	25571.9	25564.2	32.4
12	25539.5	25531.8	36.5
13	25503.0	25495.4	41.6
14	25461.4	25453.9	45.5
15	25415.9	25408.3	50.5
16	25363.4	25357.8	53.5
17	25309.9	25302.3	3.0

$$\overline{\Delta^2 V}_Q = 3.287 = 2e ; \quad e = (\Delta V - \Delta V') = 1.64.$$

from the average value of $\Delta^2 V$, can determine
e from

$$V_{P,R} = V_0 + (B_V' + B_V'')m + (B_V' - B_V'')m^2$$

$$= d = e$$

$$\overline{\Delta^2 V} = 2e = 2(B_V' - B_V'') \quad \text{like 2nd derivative}$$

$$\frac{d^2 V}{dm^2} = 2e.$$

for a ballpark estimate of $(B_V' + B_V'')$

$$\Delta V(m=+1) + \Delta V(m=-1) = 2d. = 2(B_V' + B_V'').$$

then solve for B_V' & B_V'' ball park values.

Results :

$$\overline{\Delta^2 V}_{P,R} = 3.256, (B_V' - B_V'') = -1.628$$

$$\overline{\Delta^2 V}_Q = 3.287, (B_V' - B_V'') = -1.64$$

then using $\Delta V(m=+1, m=-1) = 52 ; (B_V' + B_V'') = 26.0$

* However the $P(m=-1)$ line should be missing, used a $P(1)$
value from Broida spectrum.

From

$$(B_v' - B_v'') = -1.635$$

$$(B_v' + B_v'') = 26.0$$

$$\begin{aligned} B_v' &= 12.183 \\ B_v'' &= 13.817 \end{aligned}$$

ball park values, not accurate.

check

$$m_{\text{head}} = 7 \quad m_{\text{head}} = \frac{B_v' + B_v''}{-2(B_v' - B_v'')} \quad \text{from } \frac{dy}{dm} = 0.$$

$$m_{\text{head}} = 7.95$$

4) More accurate technique to solve for B_v' & B_v'' :

combination relations:

if have only P & R branch:

$$R(\tau-1) - R(\tau+1) = \Delta_2 F''(\tau) = 4 B_v''(\tau+\nu_2)$$

$$R(\tau) - R(\tau) = \Delta_2 F'(\tau) = 4 B_v'(\tau+\nu_2)$$

If have Q branch also :

$$\Delta_1 F' = R(\bar{J}) - Q(\bar{J})$$

$$= Q(\bar{J}+1) - P(\bar{J}+1)$$

$$\Delta_1 F'' = 2B_{\bar{J}}'(\bar{J}+1)$$

$$\Delta_1 F'' = R(\bar{J}) - Q(\bar{J}+1)$$

$$= Q(\bar{J}) - P(\bar{J}+1)$$

$$R(\bar{J}) = V_R = V_0 + 2B_{\bar{J}}' + (3B_{\bar{J}}' - B_{\bar{J}}'')\bar{J} + (B_{\bar{J}}' - B_{\bar{J}}'')\bar{J}^2$$

$$P(\bar{J}) = V_P = V_0 - (B_{\bar{J}}' + B_{\bar{J}}'')\bar{J} + (B_{\bar{J}}' - B_{\bar{J}}'')\bar{J}^2$$

$$Q(\bar{J}) = V_Q = V_0 + (B_{\bar{J}}' - B_{\bar{J}}'')\bar{J} + (B_{\bar{J}}' - B_{\bar{J}}'')\bar{J}^2$$

* even with a Q branch, the P & R relationships still hold.

$\frac{\bar{J}}{O}$	$\frac{R(\bar{J})}{25772.3}$	$\frac{Q(\bar{J})}{25737.2}$	$\frac{P(\bar{J})}{25695.9}$	$\frac{\Delta_1 F'(R, Q)}{2(\bar{J}+1)}$	$\frac{\Delta_1 F'(P, Q)}{2(\bar{J}+1)}$	$\frac{\Delta_1 F''(R, Q)}{2(\bar{J}+1)}$	$\frac{\Delta_1 F''(P, Q)}{2(\bar{J}+1)}$
1	25793.1	25737.2	25695.9	13.97	9.53	14.78	10.32
2	25805.2	25734.0	25669.6	11.86	9.5	13.1	10.73
3	25814.8	25726.6	25669.6	11.0	9.49	12.2	10.68
4	25821.2	25717.1	25641.2	10.4	9.59	11.57	10.75
5	25824.4	25705.5	25609.6	9.9	9.63	11.14	10.87
6	25826.4	25690.7	25575.1	9.7	9.37	10.97	10.65
7	25824.4	25672.8	25541.6	9.5	9.42	10.73	10.68

<u>J</u>	<u>R(J)</u>	<u>Q(J)</u>	<u>P(J)</u>	<u>$\frac{\Delta F(K,Q)}{2(J+1)}$</u>	<u>$\frac{\Delta F(P,Q)}{2(J+1)}$</u>	<u>$\frac{\Delta F(R,Q)}{2(J+1)}$</u>
1	25772.3	25737.2	(25726.3)	13.97	9.53.	14.78
2	25805.2	25734.0	25695.9	11.86	9.5.	13.1
3	25814.8	25726.6	25669.6	11.0	9.49.	12.2
4	25821.2	25717.1	25641.2	10.4	9.59.	11.57.
5	25824.4	25705.5	25609.6	9.9	9.63.	11.14
6	25826.4	25690.7	25575.1	9.7	9.37.	10.97
7	25824.4	25672.8	25541.6	9.48.	9.42.	10.73.
8	25819.1	25652.7	25502.0	9.24	9.34	10.53
9	25811.6	25629.6	25461.5	9.1	9.22	10.47
10	25798.9	25602.3	25417.9	8.94	9.15.	10.59.
11	25792.9	25571.9	25370.5	8.79	9.14	10.32
12	25761.6	25539.5	25320.2	8.54	9.04	10.14
13	25503.0	25267.9		8.95	10.49	10.54
14	25461.4	25210.9		8.92	10.43.	10.45
15	25415.9	25151.0		8.69	10.35.	10.27
16	25365.4	25087.9		8.55.	10.2	
17	25309.9	25019.1				

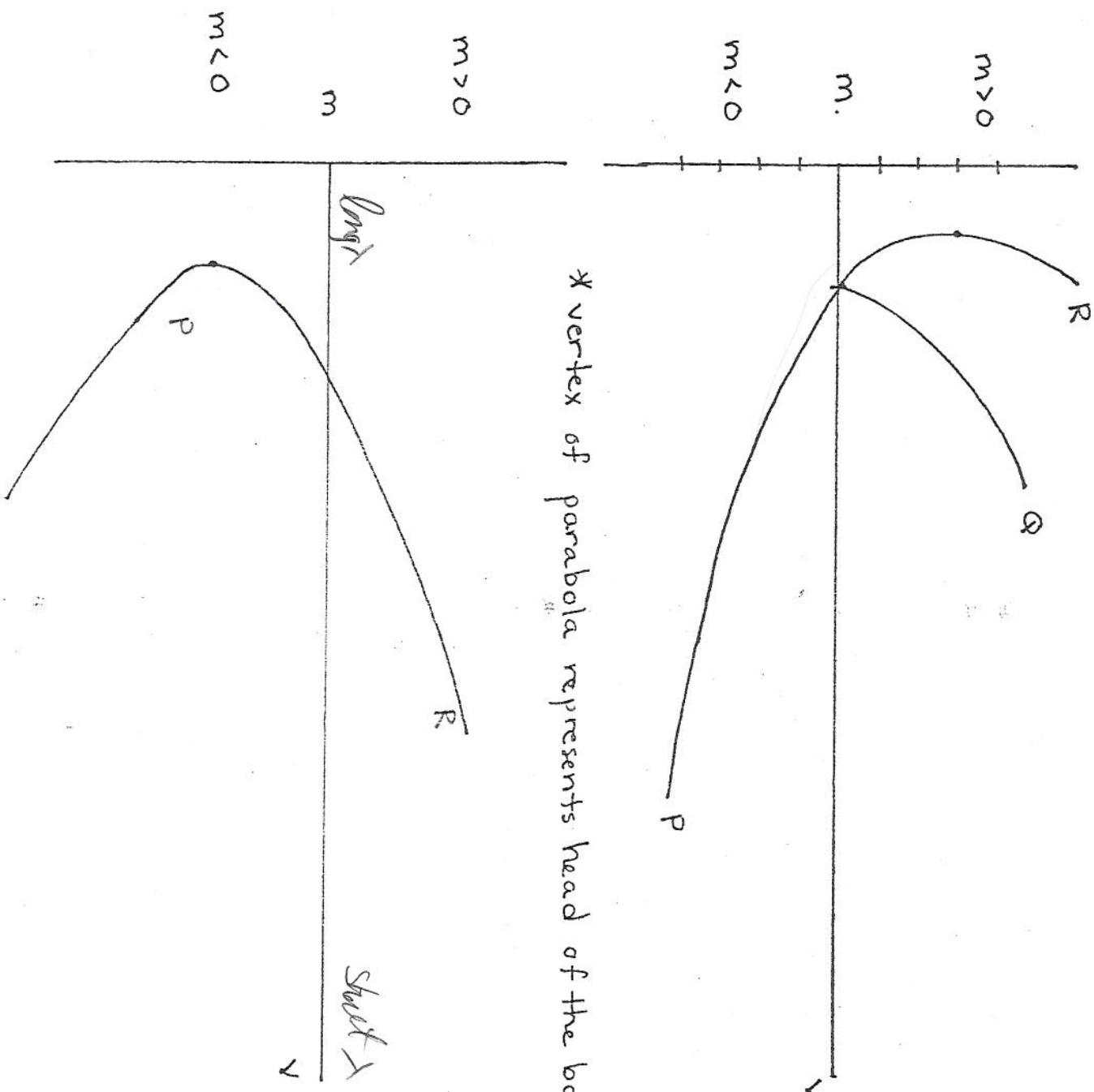
$$\langle B_v'' \rangle = 10.08$$

$$\langle B_v'' \rangle = 9.22$$

$$\langle B_v'' \rangle = 11.19$$

$$\langle B_v'' \rangle = 10.54$$

The formula for these lines represents a Parabola, called Forstrat parabola.



* vertex of parabola represents head of the band

