

Carter

2.3

a. $P = Av$

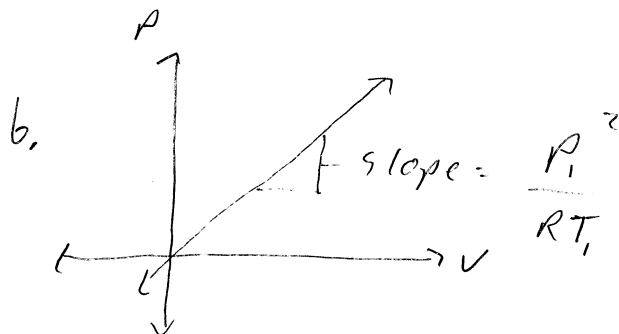
$$P_1 = Av_1$$

$$P_1 = \frac{ART_1}{P_1}$$

$$A = \frac{P_1^2}{RT_1}$$

$$P_1 v_1 = RT_1$$

$$v_1 = \frac{RT_1}{P_1}$$



c. $T = \frac{Pv}{R} = \frac{Av^2}{R}$

$$T_1 = \frac{Av_1^2}{R}$$

$$T \propto v^2; \quad v = 2v_1$$

$$T = \frac{A}{R} \cdot 4v_1^2 = 4T_1 = \boxed{800 \text{ K}}$$

A-1 a. $dz = 2x \ln y \, dx + \left(\frac{x^2}{y}\right) dy$

$$A = 2x \ln(y)$$

$$B = \frac{x^2}{y}$$

$$\frac{\partial A}{\partial y} = \frac{2x}{y}$$

$$\frac{\partial B}{\partial x} = \frac{2x}{y}$$

✓
Exact

$$\frac{\partial f}{\partial x} = A = 2x \ln y$$

$$\frac{\partial f}{\partial y} = B = \frac{x^2}{y}$$

$$f = x^2 \ln y + u(x)$$

$$f = x^2 \ln y + v(x)$$

$$\boxed{f = x^2 \ln y + C}$$

$$b. \quad dz = (y-1)dx + (x-3)dy$$

$$A = y-1$$

$$B = x-3$$

$$\frac{\partial A}{\partial y} = 1$$

$$\frac{\partial B}{\partial x} = 1 \quad \text{Exact}$$

$$\frac{\partial f}{\partial x} = A = y-1$$

$$\frac{\partial f}{\partial y} = B = x-3$$

$$f = xy - x + u(y)$$

$$f = xy - 3y + u(x)$$

$$u(y) = -3y$$

$$u(x) = -x$$

$$f = xy - x - 3y + C$$

$$c. \quad dz = (2y^3 - 3x)dx - 4xy dy$$

$$A = 2y^3 - 3x$$

$$B = -4xy$$

$$\frac{\partial A}{\partial y} = 6y^2$$

$$\frac{\partial B}{\partial x} = -4y \quad \text{Inexact.}$$

A-4

$$a. \quad dz = 2y dx + 3x dy$$

$$\frac{\partial A}{\partial y} = 2$$

$$\frac{\partial B}{\partial x} = 3 \quad \text{Inexact}$$

$$dw = 2yu dx + 3xu dy$$

Assume u is a function of x only
(see Carter pg. 397)

$$u = \exp \left[\int \frac{1}{3x} (2-3) dx \right] = \exp \left[\int -\frac{1}{3x} dx \right]$$

$$= \exp \left[-\frac{1}{3} \ln x + C \right] = \exp \left[\ln x^{-1/3} + C \right] \quad \begin{array}{l} \text{Choose} \\ C=0. \end{array}$$

$$u = x^{-1/3}$$

$$dw = 2yx^{-1/3} dx + 3x^{2/3} dy$$

$$\frac{\partial A}{\partial y} = 2x^{-1/3}$$

$$\frac{\partial B}{\partial x} = 2x^{-1/3} \quad \checkmark$$

$$b. \quad dz = y dx + (2x - y^2) dy$$

$$\frac{\partial A}{\partial y} = 1$$

$$\frac{\partial B}{\partial x} = 2 \quad \text{Inexact}$$

$$dw = yu dx + u(2x - y^2) dy$$

$$\frac{\partial}{\partial y} (yu) = \frac{\partial}{\partial x} (u(2x - y^2))$$

$$u + y \frac{du}{dy} = \frac{\partial u}{\partial x} (2x - y^2) + u \cdot 2$$

$$\text{Easier if } u = u(y)$$

$$u + y \frac{du}{dy} = 2u \quad y \frac{du}{dy} = u \quad \frac{1}{u} du = \frac{1}{y} dy$$

$$u = y$$

$$dw = y^2 dx + y(2x - y^2) dy$$

$$\frac{\partial A}{\partial y} = 2y$$

$$\frac{\partial B}{\partial x} = 2y \quad \checkmark$$

$$c. \quad dZ = (x^4 + y^4) dx - xy^3 dy$$

$$\frac{\partial A}{\partial y} = 4y^3 \quad \frac{\partial B}{\partial x} = -y^3 \quad \text{Inexact}$$

$$dw = u(x^4 + y^4) dx - uxy^3 dy$$

$$\frac{\partial}{\partial y} (u(x^4 + y^4)) = \frac{\partial}{\partial x} (-uxy^3)$$

$$\frac{\partial u}{\partial y} (x^4 + y^4) + u \cdot 4y^3 = -\frac{\partial u}{\partial x} \cdot xy^3 - uy^3$$

$$\text{Let } u = u(x)$$

$$\cancel{4y^3}u = -x\cancel{y^3} \frac{\partial u}{\partial x} - u\cancel{y^3} \quad \text{(can cancel } y^3 \text{)}$$

$$-4u = x \frac{\partial u}{\partial x} + u$$

$$x \frac{\partial u}{\partial x} = -5u \quad \frac{\partial u}{\partial u} = \frac{-dx}{x}$$

Integrating both sides:

$$\frac{1}{-5} \ln u = -\ln x$$

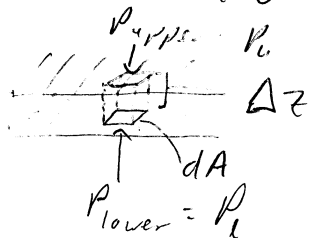
$$u = x^{-5}$$

$$dw = x^{-5}(x^4 + y^4) dx - x^{-4}y^3 dy$$

$$\frac{\partial A}{\partial y} = 4x^{-5}y^3 \quad \frac{\partial B}{\partial x} = 4x^{-5}y^3 \quad \checkmark$$

Schroeder 1.16

a.



$$P_l dA - P_u dA = \rho dA \Delta z$$

where ρ is the density of air.

$$-(P_u - P_l) = \rho dA \Delta z$$

$$-\frac{\Delta P}{\Delta z} = \rho$$

$$\left(\frac{dP}{dz} = -\rho \right)$$

b, $\rho = \frac{\text{mass}}{\text{Volume}} = \frac{n \cdot m \cdot N_A}{V}$, But $PV = nRT$.

$$\left(\text{or } \frac{P}{RT} = \frac{n}{V} \right)$$

So: $\rho = \frac{P}{RT} \cdot m \cdot N_A$

using $k = \frac{R}{N_A}$; $\frac{1}{k} = \frac{N_A}{R}$, ρ becomes

$$\rho = \frac{mP}{kT}$$

$$\left(\frac{dP}{dz} = -\frac{gmP}{kT} \right)$$

c. $\frac{1}{P} dP = -\frac{gm}{kT} dz$

Integrating both sides:

$$\ln P = -\frac{gm}{kT} z + C$$

$$P(z) = e^C \cdot e^{-mgz/kT}$$

$$P(z=0) = P_0 \Rightarrow e^C = P_0$$

$$\left(P(z) = P_0 e^{-mgz/kT} \right)$$

$$\left(\rho = \frac{mP}{kT} = \frac{mP_0}{kT} e^{-mgz/kT} \right)$$

$$d. \quad m = 0.78 \cdot 14.2 + 0.21 \cdot 16.2 + 0.01 \cdot 39.95$$

$$m = 28.96 \text{ amu} = \frac{28.96}{N_A} \text{ g} = \frac{0.02896}{N_A} \text{ kg}$$

$$P(1430) = 1 \text{ atm} \cdot \exp\left[-\frac{m g \cdot 1430 \text{ m}}{k T}\right]; g = 9.8$$

Remember, we need m in kg. $\dots$ m/s^2

$$= 1 \text{ atm} \cdot \exp\left[-\frac{0.02896 \text{ g} \cdot 1430 \text{ m}}{N_A k T}\right]$$

$$= 1 \text{ atm} \cdot \exp\left[-\frac{0.02896 \text{ g} \cdot 1430 \text{ m}}{R T}\right] \quad \left(\text{using } k = \frac{R}{N_A}\right)$$

Assume a reasonable T (I use $T = 280 \text{ K}$)

$$\frac{0.02896 \text{ g}}{R T} = 1.22 \cdot 10^{-4} \text{ m}^{-1}$$

$$P(1430) = 1 \cdot e^{-1.22 \cdot 10^{-4} \cdot 1430}$$

$$P(1430) = 0.84 \text{ atm}$$

$$P(3090) = 1 \text{ atm} \cdot e^{-1.22 \cdot 10^{-4} \cdot 3090}$$

$$= 0.686 \text{ atm}$$

$$P(4420) = 1 \text{ atm} \cdot e^{-1.22 \cdot 10^{-4} \cdot 4420}$$

$$= 0.583 \text{ atm}$$

$$P(8850) = 1 \text{ atm} \cdot e^{-1.22 \cdot 10^{-4} \cdot 8850}$$

$$= 0.34 \text{ atm}$$

Schroeder

1.17

$$a. \quad PV = nRT \left(1 + \frac{B(T)}{V/n} \right)$$

$$\frac{V}{n} = \frac{RT}{P} \left(1 + \frac{B(T)}{V/n} \right) \quad \text{For the purpose of finding } V/n, \text{ we may assume this term is small.}$$

$$\frac{V}{n} \approx \frac{RT}{P}$$

$$\frac{B(T)}{V/n} \approx \frac{B(T)}{\frac{RT}{P}} = \frac{P}{RT} \cdot B(T); \quad P = 1 \text{ atm} = 1.013 \cdot 10^5 \text{ Pa}$$

$$\frac{B(T_1)}{V/n} = \frac{P B_1}{RT_1} = \frac{1.013 \cdot 10^5 \text{ Pa} \cdot -160 \cdot 10^{-6} \text{ m}^3/\text{mol}}{8.315 \frac{\text{J}}{\text{mol} \cdot \text{K}} \cdot 100 \text{ K}}$$

$$= -0.0195$$

$$\frac{B(T_2)}{V/n} = \frac{P B_2}{RT_2} = \frac{12,183 \cdot -35 \cdot 10^{-6}}{200} = -0.0021$$

$$\frac{B(T_3)}{V/n} = \frac{P B_3}{RT_3} = \frac{12,183 \cdot -4.2 \cdot 10^{-6}}{300} = -0.00017$$

$$\frac{B(T_4)}{V/n} = \frac{P B_4}{RT_4} = 0.00027$$

$$\frac{B(T_5)}{V/n} = \frac{P B_5}{RT_5} = 0.00041$$

$$\frac{B(T_6)}{V/n} = \frac{P B_6}{RT_6} = 0.00043$$

b. Forces between molecules can be either attractive or repulsive. Attractive forces (e.g. dipole-dipole interactions, Van der Waals forces) come into play when molecules are close but not touching. However, at very close distances, repulsive forces dominate. At low T , molecules spend more time at intermediate range, so the attractive forces have a chance to work. At high T , interactions tend to happen more quickly. Molecules get close, collide, then get out. The repulsive forces dominate.

$$c. \left(P + \frac{an^2}{V^2} \right) (V - nb) = nRT$$

$$\left(PV + \frac{an^2}{V} \right) \left(1 - \frac{nb}{V} \right) = nRT$$

$$PV + \frac{an^2}{V} = nRT \left(1 - \frac{nb}{V} \right)^{-1}$$

$$\left(1 - \frac{nb}{V} \right)^{-1} \approx 1 + \frac{nb}{V} + \frac{1}{2} \cdot (-1)(-2) \left(\frac{nb}{V} \right)^2 + \dots$$

$$\approx 1 + \frac{nb}{V} + \left(\frac{nb}{V} \right)^2$$

Note that $\frac{an^2}{V} = \frac{an^2}{nRTV} \cdot nRT$,

$$PV = nRT \left[\left(1 - \frac{nb}{V} \right)^{-1} - \frac{an^2}{nRTV} \right]$$

$$\approx nRT \left[1 + \frac{nb}{V} + \left(\frac{nb}{V} \right)^2 + \dots - \frac{an^2}{nRTV} \right]$$

$$\approx nRT \left[1 + \frac{b}{(V/n)} + \frac{b^2}{(V/n)^2} + \dots - \frac{a}{RT(V/n)} \right]$$

$$\Rightarrow \left(B(T) \approx b - \frac{a}{RT} \right) \quad \left(C(T) \approx b^2 \right)$$

Additional Problems #1

$$a. \quad I = \int_{(0,0)}^{(1,1)} 2xy \, dx + x^2 \, dy$$

$$i. I = \int_0^1 2xy \, dx + \int_0^1 x^2 \, dy \quad ; y=x$$

$$\begin{aligned} I &= \int_0^1 2x^2 \, dx + \int_0^1 y^2 \, dy \\ &= \left[\frac{2x^3}{3} \right]_0^1 + \left[\frac{y^3}{3} \right]_0^1 \\ &= \frac{2}{3} + \frac{1}{3} = \boxed{1} \end{aligned}$$

$$\begin{aligned} ii. \quad I &= \int_0^1 2xy \, dx + \int_0^1 x^2 \, dy \quad ; y=x^2 \\ &= \int_0^1 2x^3 \, dx + \int_0^1 y \, dy \\ &= \left[\frac{2x^4}{4} \right]_0^1 + \left[\frac{y^2}{2} \right]_0^1 = \frac{1}{2} + \frac{1}{2} = \boxed{1} \end{aligned}$$

$$iii. \quad I = \int_0^1 2xy \, dx + \int_0^1 x^2 \, dy \quad ; y=x^3$$

$$\begin{aligned} I &= \int_0^1 2x^4 \, dx + \int_0^1 y^{2/3} \, dy \\ &= \left[\frac{2x^5}{5} \right]_0^1 + \left[\frac{3}{5} y^{5/3} \right]_0^1 \\ &= \frac{2}{5} + \frac{3}{5} = \boxed{1} \end{aligned}$$

$$\begin{aligned}
 \text{iv. } I &= \int_0^1 2xy \, dx + \int_0^1 x^2 \, dy; \quad y = \begin{cases} 0; & 0 \leq x < 1 \\ x=1 & @ \\ & x=1 \end{cases} \\
 &= \int_0^1 0 \, dx + \int_0^1 1 \, dy \\
 &= 0 + 1 = \boxed{1}
 \end{aligned}$$

$$\begin{aligned}
 \text{b. } A &= 2xy, \quad B = x^2 \\
 \frac{\partial A}{\partial y} &= 2x \quad \frac{\partial B}{\partial x} = 2x \quad \checkmark \quad \text{Exact.}
 \end{aligned}$$

$$\begin{aligned}
 \text{c. } \frac{\partial f}{\partial x} &= A = 2xy & \frac{\partial f}{\partial y} &= B = x^2 \\
 &\Rightarrow & & \\
 f &= x^2 y + u(y) & f &= x^2 y + v(x) \\
 &\text{Evidently} & & \\
 &\boxed{f = x^2 y + C} & & (C \text{ is a constant})
 \end{aligned}$$

$$\begin{aligned}
 \text{d. } f(1,1) &= 1^2 \cdot 1 + C \\
 f(0,0) &= 0 + C \\
 f(1,1) - f(0,0) &= 1 \quad \checkmark
 \end{aligned}$$

All exact differentials have this property.

Additional Problems # 2

$$V = \frac{4}{3} \pi a b c \quad \text{Dynamic Variables: } a, c$$

we have b & S constant. Since we don't have a nice expression for b and we're not differentiating with respect to it, forget it. Regard it as a constant and say

$$S = S(a, c) \leftrightarrow a = a(S, c) \leftrightarrow c = c(a, S)$$

we want

$$(1.) \left. \frac{\partial V}{\partial a} \right|_S = \frac{4}{3} \pi b \left. \frac{\partial}{\partial a} (ac) \right|_S = \frac{4}{3} \pi b \left[\underset{\uparrow}{c} + a \left. \frac{\partial c}{\partial a} \right|_S \right]$$

Using the cyclic relation:

Is a function of a

$$\left. \frac{\partial c}{\partial a} \right|_S \cdot \left. \frac{\partial a}{\partial S} \right|_c \cdot \left. \frac{\partial S}{\partial c} \right|_a = -1$$

$$\left. \frac{\partial c}{\partial a} \right|_S = \frac{-1}{\left. \frac{\partial a}{\partial S} \right|_c \cdot \left. \frac{\partial S}{\partial c} \right|_a}$$

Using the inv. relation:

$$\left. \frac{\partial a}{\partial S} \right|_c = \frac{1}{\left. \frac{\partial S}{\partial a} \right|_c}$$

$$\left. \frac{\partial c}{\partial a} \right|_S = - \frac{\left. \frac{\partial S}{\partial a} \right|_c}{\left. \frac{\partial S}{\partial c} \right|_a}$$

Since we already have a (relatively) simple expression for S in terms of a, b, c , this makes our lives easier

$$\frac{\partial S}{\partial a} \Big|_c = \frac{\partial}{\partial a} \left[4\pi \left(\frac{a^p b^p + a^p c^p + b^p c^p}{3} \right)^{1/p} \right]$$

$$= \frac{4\pi}{p} \left(\frac{a^p b^p + a^p c^p + b^p c^p}{3} \right)^{\frac{1}{p}-1} \cdot (p a^{p-1} b^p + p a^{p-1} c^p)$$

$$\frac{\partial S}{\partial c} \Big|_a = \frac{4\pi}{p} \left(\frac{a^p b^p + a^p c^p + b^p c^p}{3} \right)^{\frac{1}{p}-1} \cdot (p a^p c^{p-1} + p b^p c^{p-1})$$

$$\frac{\partial c}{\partial a} \Big|_s = - \frac{\frac{\partial S}{\partial a} \Big|_c}{\frac{\partial S}{\partial c} \Big|_a} = - \frac{(p a^{p-1} b^p + p a^{p-1} c^p)}{(p a^p c^{p-1} + p b^p c^{p-1})} = \frac{-a^{p-1}(b^p + c^p)}{c^{p-1}(a^p + b^p)}$$

Getting c as a function of a is merely algebra:

$$S = 4\pi \left(\frac{a^p b^p + a^p c^p + b^p c^p}{3} \right)^{1/p}$$

$$3 \left(\frac{S}{4\pi} \right)^p - a^p b^p = c^p (a^p + b^p)$$

$$c = \left(\frac{3 \left(\frac{S}{4\pi} \right)^p - a^p b^p}{a^p + b^p} \right)^{1/p}$$

At this point, simply substitute c & $\frac{\partial c}{\partial a} \Big|_s$ into (1.)