

$$1) \quad \boxed{S' = S - \frac{1}{T} U}$$

$$dS' = dS - d\left(\frac{1}{T}\right) U - \frac{1}{T} dU$$

$$\boxed{dS' = -U d\left(\frac{1}{T}\right) + \frac{P}{T} dV - \frac{\mu}{T} dN}$$

$$\boxed{S'' = S - \frac{1}{T} U - \frac{P}{T} V}$$

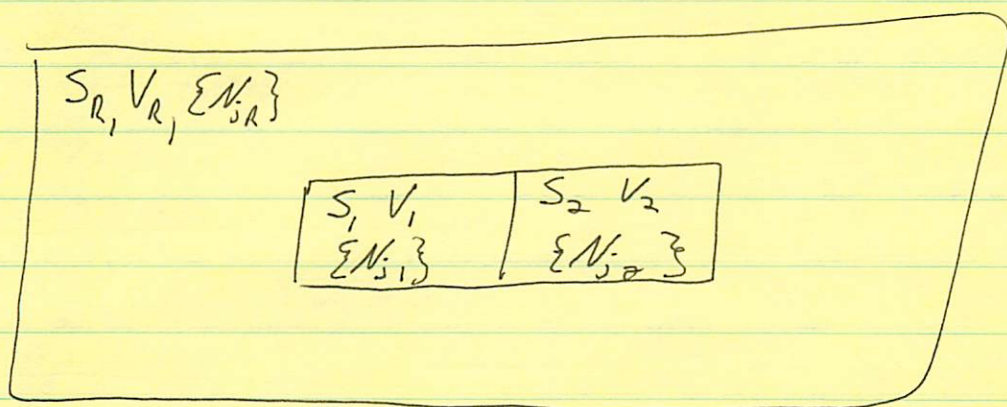
$$dS'' = dS - U d\left(\frac{1}{T}\right) - \frac{1}{T} dU - V d\left(\frac{P}{T}\right) - \frac{P}{T} dV$$

$$\boxed{dS'' = -U d\left(\frac{1}{T}\right) - V d\left(\frac{P}{T}\right) - \frac{\mu}{T} dN}$$

$$\boxed{S''' = S - \frac{1}{T} U + \frac{\mu}{T} N}$$

$$dS''' = -U d\left(\frac{1}{T}\right) + \frac{P}{T} dV + ~~\frac{\mu}{T} dN~~ N d\left(\frac{\mu}{T}\right)$$

2) Consider a composite system composed of a large reservoir and a subsystem with internal variables as in class. To invoke therm. min principle, we must assume $S_{TOT}, V_{TOT}, N_{TOT}$ are all constant



We want the 1 & 2 subsystem to be maintained at constant $\{U, V, T\}$ through interaction with the reservoir. This means S_1, S_2, S_R can all vary subject to $S_1 + S_2 + S_R = \text{constant}$

V_1, V_2 can vary subject to $V_1 + V_2 = \text{constant}$

N_{j1}, N_{j2} can vary subject to $N_{j1} + N_{j2} = \text{constant}$

$dU_{TOT} = 0$ at equilibrium

$$dU_{TOT} = dU_R + dU_1 + dU_2$$

$$dU_R = \left. \frac{\partial U_R}{\partial S_R} \right|_{V_R, \{N_{jR}\}} dS_R = T_R dS_R$$

but, $dS_R = -dS_1 - dS_2$ since $S_R + S_1 + S_2 = \text{const}$

so,
$$\begin{aligned} dU_R &= dU_1 + dU_2 - T_R dS_1 - T_R dS_2 \\ &= d(U_1 - T_R S_1) + d(U_2 - T_R S_2) \\ &= dA_1 + dA_2 \\ &= dA_{\text{sys}} \end{aligned}$$

Reservoir is large, so T_R unaffected by changes in system.

i.e. $dA_{\text{sys}} = 0$ at equilibrium for

$V_{\text{sys}} \text{ constant, } \{N_j\}_{\text{sys}} \text{ constant}$

$T_{\text{sys}} = T_R \text{ constant}$